

Role of Multi-Electron Effects in the Asymmetry of Strong-Field Ionization and Fragmentation of Polar Molecules: The Methyl Halide Series

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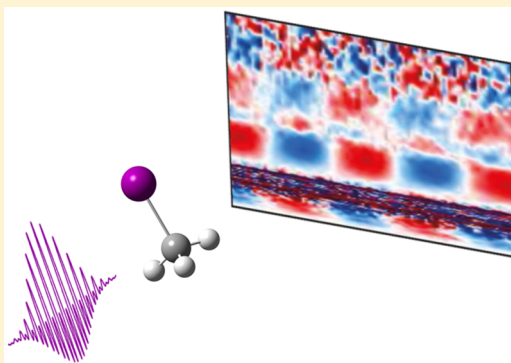
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ABSTRACT: We report angle- and momentum-resolved measurements of the dissociative ionization and Coulomb explosion of methyl halides (CH_3F , CH_3Cl , CH_3Br , and CH_3I) in intense phase-controlled two-color laser fields. At moderate laser intensities, we find that the emission asymmetry of low-energy CH_3^+ fragments from the $\text{CH}_3^+ + \text{X}^+$ ($\text{X} = \text{F}$, Cl , Br , or I) channel reflects the asymmetry of the highest occupied molecular orbital of the neutral molecule with important contributions from the Stark effect. This asymmetry is correctly predicted by the weak-field asymptotic theory, provided that the Stark effect on the ionization potentials is calculated using a nonperturbative multielectron approach. In the case of high laser intensities, we observe a reversal of the emission asymmetries for high-energy CH_3^+ fragments, originating from the dissociation of CH_3X^{q+} with $q \geq 2$. We propose ionization to electronically excited states to be at the origin of the reversed asymmetries. We also report the measurements of the emission asymmetry of H_3^+ , which is found to be identical to that of the low-energy CH_3^+ fragments measured at moderate laser intensities. All observed fragmentation channels are assigned with the help of CCSD(T) calculations. Our results provide a benchmark for theories of strong-field processes and demonstrate the importance of multielectron effects in new aspects of the molecular response to intense laser fields.



INTRODUCTION

The ionization of molecules in strong laser fields is the common first step of several new techniques for probing the dynamics of molecules. Understanding the ionization process is therefore essential for the further development of high-harmonic spectroscopy,^{1–6} laser-induced electron diffraction,^{7–10} and strong-field photoelectron holography.^{11–13} However, accurate theories are still rare and polar molecules have proven particularly challenging to describe. According to the most elementary theories, such as the molecular Ammosov-Delone-Krainov theory (MO-ADK)¹⁴ or the strong-field approximation (SFA),¹⁵ the tunneling ionization rate of a molecule is expected to maximize when the electron is removed via the location of maximal density of the highest-occupied molecular orbital (HOMO). Improvements over these basic theories must take into account the modification of the ionization potential by the Stark effect. This is crucial for the methyl halides which all have large permanent dipole moments (CH_3F : 1.85D, CH_3Cl : 1.87D, CH_3Br : 1.81D, and CH_3I : 1.62D).¹⁶ In addition, particular attention must be paid to the accuracy of the asymptotic part of the orbital wave functions which is crucial for obtaining accurate ionization rates. Finally, parabolic coordinates naturally provide a more accurate

description of strong-field ionization in the tunneling limit because of the near-separability of coordinates. These essential developments have been implemented in the weak-field asymptotic theory (WFAT).^{17–20} Other methods that have proven successful in describing strong-field ionization (SFI) of polar molecules include time-dependent multielectron methods,^{21–25} but the predictive power of all theories largely remains to be established.

The challenge posed by SFI of polar molecules is well-illustrated by the example of the carbonyl sulfide (OCS) molecule. This molecule has been examined by several research groups using different methods.^{26–31} In accordance with the most elementary theories, the SFI rate is expected to be largest when the electron is removed via the S atom where the density of the HOMO is maximal. However, Holmegaard et al.²⁶ showed experimentally that for circularly polarized laser pulses with a peak intensity of $\approx 2.4 \times 10^{14} \text{ W/cm}^2$, the ionization of OCS is enhanced when the field points from the O to the S atom. They showed that the discrepancy between the calculated

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and observed ionization asymmetries may be attributed to the neglected Stark shift of the ionization potential in MO-ADK theory. Recently Ohmura et al.³¹ presented experimental results where OCS was ionized by linearly polarized phase-controlled two-color pulses with a peak intensity of 5×10^{13} W/cm². The two-color experiment suggested that ionization is enhanced when the electron is removed via the S atom. This result agreed with the ionization asymmetry predicted by the WFAT.³² The example of the OCS molecule illustrates the fact that the electron density of the HOMO and the associated linear Stark effect act in general antagonistically on the asymmetry of SFI. As a consequence, the accurate calculation of SFI rates is a challenging unresolved task, especially when multielectron effects are important.

In this paper, we report measurements of the asymmetries of dissociative ionization and Coulomb explosion of the methyl halides (CH₃X, with X = F, Cl, Br, or I) in phase-controlled two-color laser fields. We measure the angle-resolved momentum distribution spectra of different charged fragments as a function of the two-color phase. The measurements were repeated for multiple laser intensities. Further, we have calculated the orientation-dependent ionization rates of the methyl halides based on the WFAT and compare its prediction with the experimental observations. We compare the results of the single-active-electron version of the WFAT, where dipole moments are calculated as expectation values over field-free orbitals, with an improved version that uses dipole moments calculated with multielectron quantum chemical methods with applied static electric fields. We show that only the latter version agrees with the experimental results. Strong-field ionization of methyl halides in a phase-controlled two-color field has previously been measured by Ohmura³³ but was restricted to time-of-flight detection and a single laser intensity. Our results confirm their basic observations but show the presence of multiple dissociative ionization and Coulomb explosion channels. Most importantly, we show that the asymmetry of a given fragment can reverse as a function of the kinetic energy release and intensity. The former observation has previously been made in two-color experiments on apolar molecules^{34–36} but not with polar molecules.^{31,37–39} It shows the necessity of performing measurements with sufficient momentum resolution to identify multiple channels and to repeat measurements with multiple intensities to reveal the complexity of SFI of polar molecules. Our results additionally serve as a benchmark for theories of molecular SFI and highlight the importance of multielectron effects in the Stark shifts.

EXPERIMENT

The experiment was carried out using a velocity-map-imaging spectrometer (VMIS).⁴⁰ The two-color pulse was synthesized using second-harmonic generation in a β -barium-borate (BBO) crystal (300 μ m thickness). A 50 fs linearly polarized pulse with a center wavelength of 800 nm passed through the crystal leading to the generation of an additional 400 nm pulse with polarization orthogonal to the incident beam. The 400 nm component had a spectral bandwidth of 6 nm, corresponding to a Fourier limit of 38 fs. A zeroth-order half-wave plate (at 800 nm) placed behind the crystal rotated the polarization plane of the 800 nm pulse by 90° to lie in the polarization plane of the 400 nm pulse. The intensity of the two-color pulse was adjusted by an iris placed directly in front of the entrance window (sapphire, 2.6 mm thickness) of the VMIS. The aperture

diameter of the iris was varied between 1.8 and 2.6 mm. A spherical mirror ($f = 100$ mm) mounted inside the VMIS focused the beam backward into the center of the supersonic gas jet. The latter was formed by a pulsed Even-Lavie valve (orifice 150 μ m, 1 kHz repetition rate). Measurements were performed with CH₃F, CH₃Cl, CH₃Br, and CH₃I. All target molecules were diluted in helium to a concentration of $\leq 1\%$ to minimize clustering and space-charge effects. The group-delay dispersion of the BBO, the half-wave plate, and the entrance window were compensated by a calcite plate (2.6 mm thickness) and two UV fused silica plates (total thickness of 2.5 mm). One UV-fused silica plate (1.5 mm thickness) was mounted on a motorized rotation stage (Newport PRS0PP). By tilting this plate, the relative phase ϕ between the ω and 2ω fields was adjusted. The intensity of the 400 nm pulse was ($22 \pm 3\%$) of the total intensity for the used range of iris aperture diameters. To obtain the momentum-distribution spectra, the measured VMIS images were Abel inverted using an iterative inversion method⁴¹ (see Figure 1). The inversion can lead to

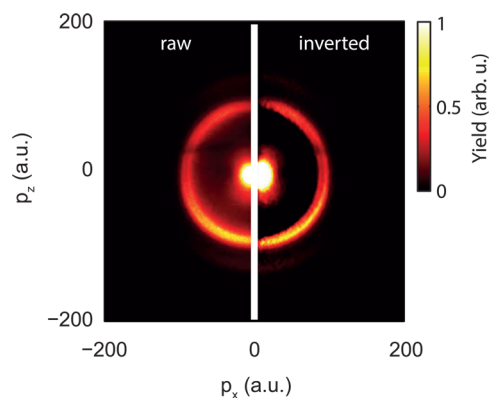


Figure 1. Raw and inverted images of CH₃⁺ from the interaction of CH₃I with an intense femtosecond 800 + 400 nm laser pulse (a.u. = atomic units). The direction of polarization of the two-color laser field was parallel to p_z .

negative numbers especially for momenta where the signal is close to zero or which are close to the symmetry axis of the spectrum. These momenta were excluded from further analysis. The spectra were then convoluted with a Gaussian filter to reduce the high-frequency noise caused by the inversion. The full width at half-maximum of the filter function was 1% of the momentum detection range. The relative phase ϕ of the two-color field defined as $E(t) = E_1(t) \cos(\omega t) + E_2(t) \cos(2\omega t + \phi)$ was calibrated by measuring the dissociative ionization of CO.³⁷

RESULTS

The left panels of Figure 2 (panels a–d) show the yield spectra of the CH₃⁺ fragments from CH₃X (where X = F, Cl, Br, or I) as a function of the kinetic energy. The CH₃⁺ ions can arise through the fragmentation of the parent ion following the reaction



where $q = 0, 1, \dots$. In the case of a two-body fragmentation process, it is sufficient to know the energy of one fragment to obtain the center-of-mass energy. Therefore, one can distinguish different dissociation channels based on the energy of the CH₃⁺ ions alone. In all spectra, local maxima or shoulders near 0 eV and 3–5 eV can be observed. The peak near 5 eV in

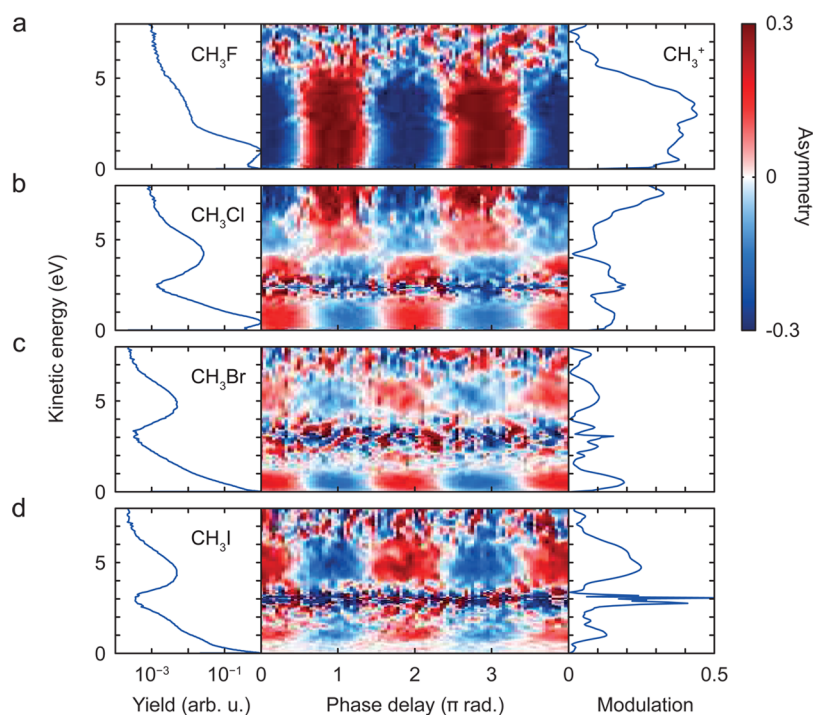


Figure 2. Asymmetry of CH_3^+ emission from methyl halides exposed to an intense two-color field. (a) Spectrum of the normalized yield (left panel), density plot of the asymmetry α defined in eq 4 (center panel) and modulation amplitude (right panel) of CH_3^+ fragments from CH_3F . The laser peak intensity was $(1.0 \pm 0.4) \times 10^{14} \text{ W/cm}^2$. (b, c, and d) Similar to (a) but from (b) CH_3Cl , (c) CH_3Br exposed to a laser peak intensity of $(5 \pm 2) \times 10^{13} \text{ W/cm}^2$, and (d) CH_3I exposed to a laser peak intensity of $(3 \pm 1) \times 10^{13} \text{ W/cm}^2$.

the yield spectrum of CH_3I is assigned to the Coulomb explosion of the doubly charged parent ion to CH_3^+ and I^+ .⁴² The peak near 0 eV is attributed to the dissociative ionization of CH_3I into CH_3^+ and I .⁴³ Since the yield spectra of all four molecules are qualitatively very similar, this assignment is most likely also valid for the other three methyl halides. This assignment was confirmed by calculating the kinetic energy of CH_3^+ resulting from the Coulomb explosion of the doubly charged parent ions using either the purely Coulombic repulsion energy or a high-level ab initio calculation. We first discuss the simple Coulomb model and then describe the ab initio calculations.

Coulomb Model. The purely Coulombic part of the kinetic energy release (E_{CM}^{C} in Tables 1–3) was calculated by assuming

Table 1. Calculated Kinetic-Energy Release (in eV) for the Dissociation Reaction $\text{CH}_3\text{X}^{2+} \rightarrow \text{CH}_3^+ + \text{X}^+$ Based On the Purely Coulombic Energies (eq 2) or An Ab-Initio Calculation Described in the Text. Both the Total Center-of-Mass Energy E_{CM}^{C} of the Two Fragment Ions and the Kinetic Energy $E_{\text{CH}_3^+}$ of the CH_3^+ Fragments are Given^a

	R_{CX}	Coulomb		Ab initio		our exp
		E_{CM}^{C}	$E_{\text{CH}_3^+}^{\text{C}}$	$E_{\text{CM}}^{\text{ai}}$	$E_{\text{CH}_3^+}^{\text{ai}}$	
CH_3F	1.46	9.9	5.5	4.2	2.3	3.4
$\text{CH}_3^{35}\text{Cl}$	1.86	7.7	5.4	5.0	3.5	4.3
$\text{CH}_3^{79}\text{Br}$	2.00	7.2	6.1	4.7	4.0	5.1
CH_3I	2.20	6.5	5.9	3.8	3.4	4.7

^aThe internuclear distance between the halogen atom X and the center of mass of the methyl group CH_3 [R_{CX} (Å)] used in the calculations and the experimental kinetic energy (KE) peak are given.

two point charges $Q_1 = pe$ and $Q_2 = qe$ to be separated by a distance R between the centers of mass of the fragments, i.e.

$$E_{\text{CM}}^{\text{C}} = \frac{pqe^2}{4\pi\epsilon_0 R} \quad (2)$$

where e is the elementary charge and ϵ_0 the permittivity of vacuum. The kinetic energy of a single fragment from a two-body dissociation is

$$E_{\text{frag}} = \frac{m_{\text{mol}} - m_{\text{frag}}}{m_{\text{mol}}} E_{\text{CM}} \quad (3)$$

where m_{mol} and m_{frag} represent the total mass of the molecule and the mass of the fragment, respectively. The results for the Coulomb explosion leading to $\text{CH}_3^+ + \text{X}^+$ are summarized in Table 1. The kinetic energy $E_{\text{CH}_3^+}^{\text{C}}$ was obtained by choosing $R = R_{\text{CX}}$ where R_{CX} represents the distance between the halogen atom and the center of mass of the methyl group. The center-of-mass energy E_{CM}^{C} decreases from CH_3F to CH_3I ; however, the kinetic energy $E_{\text{CH}_3^+}^{\text{C}}$ of the CH_3^+ fragments lies between 5.4 and 6.1 eV for all molecules. The obtained energies are systematically higher than the positions of the observed peaks in the kinetic-energy spectra.

Ab Initio Calculations. Much more accurate values for the kinetic-energy release were obtained from ab initio quantum chemical calculations. We used the CCSD(T) method with the cc-pVQZ and cc-pVSZ all-electron basis sets for X = F, Cl, and Br and corresponding pseudopotential basis sets for X = Br and I. Electronic energies were extrapolated to the complete basis set limit. Relativistic effects have been estimated with the second-order Douglas-Kroll-Hess (DKH2) and fourth-order Douglas-Kroll-Hess (DKH4) approximations using the cc-pVSZ-DK basis sets for all electron calculations for X = F, Cl,

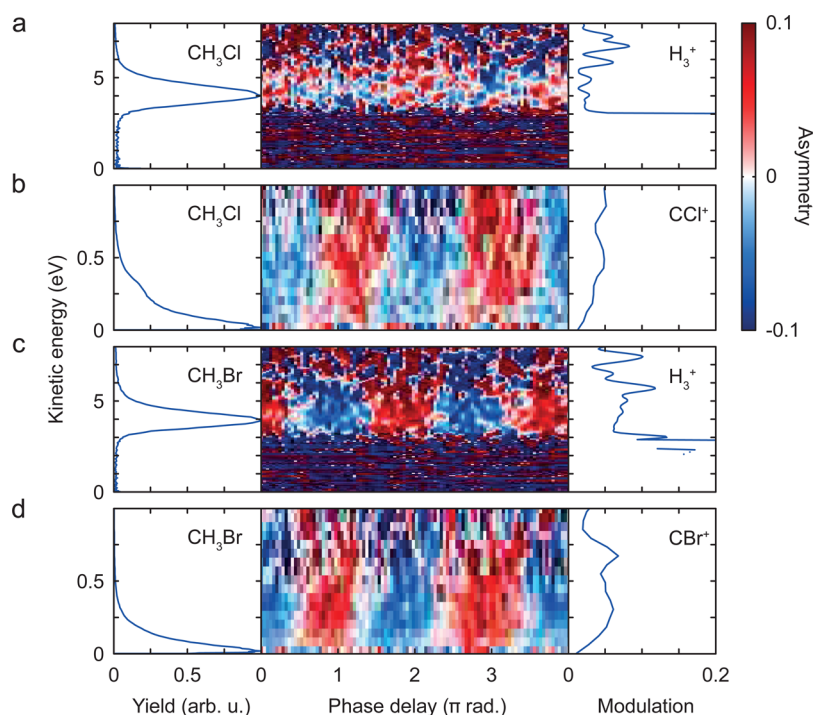


Figure 3. Dissociative ionization leading to the formation of H_3^+ . Similar to Figure 2 but for (a) H_3^+ and (b) CCl^+ from CH_3Cl and (c) H_3^+ and (d) CBr^+ from CH_3Br . The laser peak intensity was $(1.6 \pm 0.6) \times 10^{14} \text{ W/cm}^2$.

and Br. These calculations reproduced the second ionization potentials of F, Cl, Br, and I within 0.4 eV. The kinetic-energy releases were calculated using MP2/cc-pVTZ-optimized geometries for CH_3X , CX^+ , CX^{2+} , CH_3^+ , and H_3^+ , assuming vertical ionization of the CH_3X molecule followed by dissociation to the equilibrium geometry of the fragments in their electronic ground state which are the following: CH_3X^{2+} ($^3\text{A}_1$, C_{3v} symmetry), CH_3X^{3+} (^2E , C_{3v} symmetry), CH_3^+ ($^1\text{A}_1$, D_{3h} symmetry), H_3^+ ($^1\text{A}_1$, D_{3h} symmetry), CX^+ ($^1\Sigma_g^+$), CX^{2+} ($^2\Sigma_g^+$), X^+ (^3P), and X^{2+} (^4S). Total predicted kinetic-energy releases ($E_{\text{CM}}^{\text{ai}}$ in Tables 1–3) were obtained from the basis set extrapolated electronic energies and including zero-point vibrational energies estimated from MP2/cc-pVTZ harmonic frequencies of the molecular fragments. Remaining basis set errors are likely to be less than 0.1 eV for the all-electron results, whereas the pseudopotential approximation appears to have errors around 0.2 eV for the triply charged species. On the basis of the MP2, CCSD, and CCSD(T) results, remaining errors in the electronic correlation energies are likely to lie below 0.2–0.3 eV. The estimated errors in the zero-point vibrational energies are in the range of 0.01 eV.

Comparison between Coulomb and Ab Initio Calculations. The kinetic energy releases $E_{\text{CM}}^{\text{ai}}$ calculated ab initio systematically lie below the purely Coulombic repulsion energies E_{CM}^{C} as a consequence of a large covalent bonding character of the electronic ground states of the doubly and triply charged methyl-halide molecules. Since the electronic ground states of these species usually have the highest covalent bond strengths, $E_{\text{CM}}^{\text{ai}}$ can be considered to represent the lower limit of the kinetic energy release from a given charge state. The kinetic energy releases from electronically excited states will in general be higher than $E_{\text{CM}}^{\text{ai}}$ and can even lie above E_{CM}^{C} when an electronic state with a repulsive covalent character is accessed. These results allow us to identify the local maxima between 3 and 7 eV in the CH_3^+ spectra of Figure 2 (panels b–

d), as well as the broad shoulder between 3 and 5 eV in Figure 2a, with CH_3^+ from the Coulomb explosion of CH_3X^{2+} . This assignment also agrees with the results of ref 42 on CH_3I^{2+} .

Asymmetry of CH_3^+ from CH_3X . The measured emission asymmetry of CH_3^+ as a function of the kinetic energy and the two-color phase ϕ is shown on the right-hand side of Figure 2 (panels a–d). The asymmetry $\alpha(E, \phi)$ as a function of the kinetic energy E and the two-color phase ϕ is defined as

$$\alpha(E, \phi) = \frac{Y_{\text{up}}(E, \phi) - Y_{\text{down}}(E, \phi)}{Y_{\text{up}}(E, \phi) + Y_{\text{down}}(E, \phi)} \quad (4)$$

where $Y_{\text{up}}(E, \phi)$ and $Y_{\text{down}}(E, \phi)$ are the ion yields collected in the upper or lower hemisphere along the direction of the polarization axis, respectively. At $\phi = 0$, the maximal amplitude of the asymmetric two-color field points in the direction called “up”. For CH_3F (Figure 2a), the asymmetry $\alpha(E, \phi)$ at $\phi = 0$ is negative over the complete energy range. Moreover, the asymmetry modulates in phase with respect to ϕ . The amplitude of the modulation as a function of the kinetic energy is shown on the right-hand side of the density plot. In the case of CH_3F , amplitudes of up to 0.44 were observed. From the emission asymmetry of CH_3^+ from CH_3F , one can conclude that both the dissociative ionization [$q = 0$ in reaction (1)] and Coulomb explosion ($q = 1$) are faster when the maximal electric field vector points from the methyl group to the fluorine atom (i.e., the electron is removed via the methyl group). For CH_3^+ emitted from CH_3Cl , CH_3Br , and CH_3I in contrast, a positive asymmetry of fragments with energies < 4 eV was observed for $\phi = 0$ (Figure 2, panels b and c). It follows that the dissociative ionization and Coulomb explosion generating the lowest-energy fragments of these molecules are enhanced when the electric field maximum points from the halogen atom to the methyl group (i.e., in the opposite direction compared to CH_3F). However, in the density plot of

CH_3Cl (Figure 2b), a clear change of the asymmetry can be observed at 4.2 eV. The CH_3^+ fragments with kinetic energies larger than 4.2 eV must therefore originate from a different channel than those with a smaller kinetic energy. The change in asymmetry occurs at the maximum of the peak assigned to the $\text{CH}_3\text{Cl}^{2+} \rightarrow \text{CH}_3^+ + \text{Cl}^+$ reaction. As we further discuss below, the reversal in the asymmetry is likely to originate from ionization to electronically excited states of $\text{CH}_3\text{Cl}^{2+}$ (e.g., by removing the first electron from HOMO and the second one from HOMO–1 or HOMO–2).

Formation and Asymmetry of H_3^+ . In addition to the previously observed dissociative ionization and Coulomb explosion channels, we have also observed the reaction



leading to the formation of H_3^+ . This reaction is particularly interesting because it requires the breaking and formation of three bonds.^{44,45} The measured asymmetries of H_3^+ and CX^+ from CH_3Cl and CH_3Br are shown in Figure 3 (panels a–d, respectively). At $\phi = 0$, the asymmetry is positive for H_3^+ from CH_3Cl and CH_3Br , identical to the asymmetry of the low-energy CH_3^+ fragments from the corresponding molecules. As expected, the asymmetry of the CX^+ fragments is opposite to the asymmetry of the H_3^+ ions. Sun et al.⁴⁴ studied the formation of H_3^+ from CH_3Cl in a symmetric femtosecond laser field and reported a maximum of the H_3^+ yield at a kinetic energy of 1.69 eV, whereas we measured an energy of 4.0 eV. The reported laser peak intensity in ref 44 was 9.4×10^{13} W/cm², not sufficiently different from the peak intensity used in our study [$(1.6 \pm 0.6) \times 10^{14}$ W/cm²] to explain the large difference in kinetic energies. Our experimental value is also closer to the predicted kinetic energy $E_{\text{H}_3^+}^{\text{ai}}$ of the H_3^+ fragments given in Table 2.

Table 2. Calculated Kinetic-Energy Release (in eV) for the Dissociation Reaction $\text{CH}_3\text{X}^{2+} \rightarrow \text{CX}^+ + \text{H}_3^+$ Based on the Purely Coulombic Energies (eq 2) or An Ab Initio Calculation Described in the Text^a

	Coulomb		ab initio	
	E_{CM}^{C}	$E_{\text{H}_3^+}^{\text{C}}$	$E_{\text{CM}}^{\text{ai}}$	$E_{\text{H}_3^+}^{\text{ai}}$
$\text{CH}_3^{35}\text{Cl}$	8.4	7.9	5.87	5.52
$\text{CH}_3^{79}\text{Br}$	7.2	6.9	4.02	3.89

^aBoth the total center-of-mass energy E_{CM} of the two fragment ions and the kinetic energy $E_{\text{H}_3^+}$ of the H_3^+ fragments are given.

Asymmetry of CH_3^+ and I^{q+} from CH_3I . We now show that the relatively simple results displayed in Figure 2 become significantly more complex at higher intensities of the two-color probe field. Here, we will mainly focus on the results obtained for CH_3I , although similar observations have also been made for CH_3Cl and CH_3Br . The effect of the laser intensity on the yield spectrum and the asymmetry of CH_3^+ from CH_3I is shown in Figure 4 (panels a–d). The measured abundance ratio is $\text{I}^+:\text{I}^{2+}:\text{I}^{3+}:\text{I}^{4+} = 1:0.47:0.10:0.01$ at the intensity of 1×10^{14} W/cm² (panel c). The yield of high-kinetic-energy fragments is found to increase as a function of intensity with the appearance of additional maxima. The evolution of the asymmetry reveals two main changes as a function of intensity. First, the emission asymmetry of the peak near 0 eV changes sign with increasing intensity. This shows that the asymmetry of the dissociative-ionization channel producing $\text{CH}_3^+ + \text{I}$ with

very small kinetic energies changes sign as a function of intensity. The second change is the appearance of high-energy CH_3^+ fragments which have the opposite asymmetry compared to the fragments observed in Figure 2d. Clear reversals of the sign of the asymmetry are observed at 6.9, 8.0, and 11.2 eV in Figure 4 (panels c and d). Deviations between the asymmetry maxima in different panels are small and within the experimental uncertainty. We note that this observation stands in contrast to results that we have obtained on CH_3Cl (not shown), where asymmetry features of Cl^+ and Cl^{2+} display continuous transitions in their phase dependence, which suggests the presence of multiple overlapping channels.

The relation between the energy $E_{\text{CH}_3^+}$ of CH_3^+ ions and the energy $E_{\text{I}^{q+}}$ of I^{q+} ions from Coulomb explosion according to eq 1 is given by

$$\frac{E_{\text{I}^{q+}}}{E_{\text{CH}_3^+}} = \frac{m_{\text{CH}_3^+}}{m_{\text{I}^{q+}}} \quad (6)$$

where $m_{\text{CH}_3^+}$ and $m_{\text{I}^{q+}}$ are the masses of the fragments. The predicted kinetic energies for CH_3^+ and I^{q+} fragments calculated using eqs 2 and 3 are given in Table 3 for $q = 1-4$. In addition, we also report the corresponding values obtained from the ab initio calculations described above. On the basis of the relation in eq 6 and together with the emission asymmetry, it is thus possible to find correlations between measurements of the CH_3^+ and I^{q+} ions. We identify the kinetic energies of CH_3^+ fragments corresponding to 4.7, 7.4, 9.8, and 12.6 eV, where we see a strongly modulating yield (see Figure 4c) and discuss the correlations with I^{q+} . These fragment kinetic energies correspond to a total kinetic-energy release (KER) of 5.3, 8.3, 11, and 14 eV, respectively. The emission asymmetries of the different ions are compared in Figure 5, (a) CH_3^+ , (b) I^+ , (c) I^{2+} , (d) I^{3+} , and (e) I^{4+} . The KER distribution is plotted on the left for comparison. The horizontal green lines are drawn at the above-mentioned KER values. The comparison and correlation between fragments based on KER is only valid for two-body dissociation channels.

(1) KER 5.3 eV: This KER corresponds to fragment kinetic energies of 4.7 and 0.6 eV for CH_3^+ and I^{q+} , respectively. Following the horizontal green line at 5.3 eV in Figure 5 (panels a and b), one finds opposite emission asymmetry for CH_3^+ and I^+ at this energy, as expected for a two-body fragmentation pathway. This KER lies between the $E_{\text{CH}_3^+}$ obtained from Coulomb and ab initio calculations in Table 3 ($q = 1$). We therefore assign the fragmentation channel to be $\text{CH}_3\text{I}^{2+} \rightarrow \text{CH}_3^+ + \text{I}^+$.

(2) KER 8.3 eV: The individual kinetic energies of CH_3^+ and I^{q+} are 7.4 and 0.9 eV, respectively, in a two-body fragmentation scheme. Following the horizontal green line at 8.3 eV from panel (a) to (b) does not show the expected reversal of asymmetry. However, the asymmetry plot of I^{2+} (c) does reveal the expected opposite asymmetry. Comparison with Coulomb and ab initio calculations for $q = 2$ in Table 3, however, shows that the KER is at least 2.7 eV smaller than expected for the $\text{CH}_3\text{I}^{3+} \rightarrow \text{CH}_3^+ + \text{I}^{2+}$ channel. Therefore, we exclude this assignment. We conclude that the CH_3^+ fragments with kinetic energies close to 7.4 eV originate from a $\text{CH}_3\text{I}^{2+} \rightarrow \text{CH}_3^+ + \text{I}^+$ channel which displays an asymmetry opposite to that of the dominant channel producing CH_3^+ with a kinetic energy centered at 4.7 eV. Ionization to electronically excited states of CH_3I^{2+} by, for example, removing one electron from HOMO

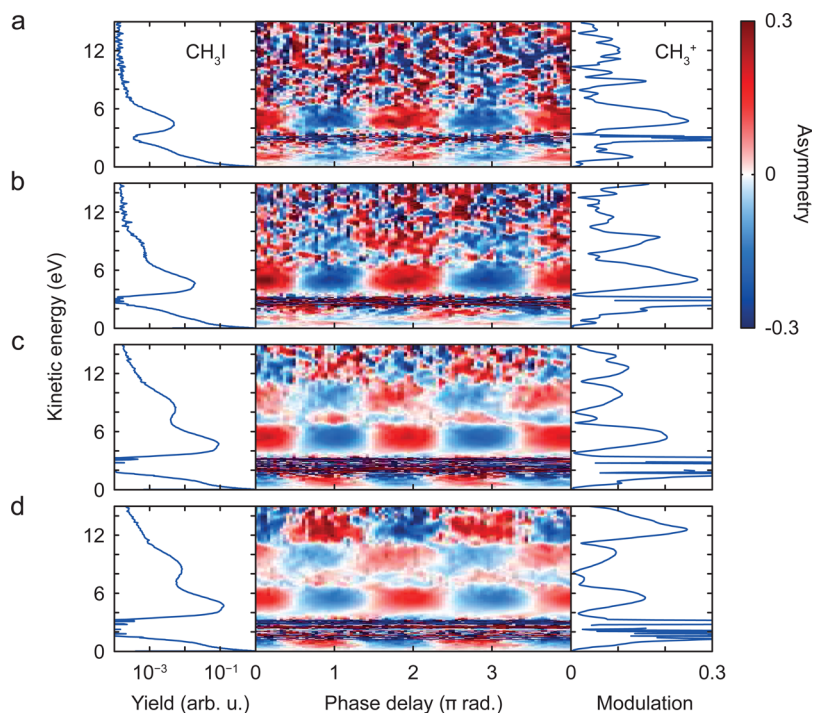


Figure 4. Intensity dependence of the dissociation asymmetry of CH_3^+ from CH_3I . The laser peak intensity was (a) $(3 \pm 1) \times 10^{13} \text{ W/cm}^2$, (b) $(5 \pm 2) \times 10^{13} \text{ W/cm}^2$, (c) $(1 \pm 0.4) \times 10^{14} \text{ W/cm}^2$, and (d) $(1.6 \pm 0.6) \times 10^{14} \text{ W/cm}^2$.

Table 3. Calculated Kinetic-Energy Release (in eV) for the Dissociation Reaction $\text{CH}_3\text{I}^{(1+q)+} \rightarrow \text{CH}_3^+ + \text{I}^{q+}$ Based On the Purely Coulombic Energies (eq 2) or An Ab Initio Calculation Described in the Text^a

q	Coulomb			ab initio		
	E_{CM}^{C}	$E_{\text{CH}_3^+}^{\text{C}}$	$E_{\text{I}^{q+}}^{\text{C}}$	$E_{\text{CM}}^{\text{ai}}$	$E_{\text{CH}_3^+}^{\text{ai}}$	$E_{\text{I}^{q+}}^{\text{ai}}$
1	6.5	5.9	0.7	3.8	3.4	0.4
2	13.1	11.7	1.4	11.3	10.1	1.2
3	19.6	17.6	2.1	14.3	12.8	1.5
4	26.2	23.4	2.8	17.4	15.6	1.8

^aBoth the total center-of-mass energy E_{CM} of the two fragment ions and the kinetic energy $E_{\text{CH}_3^+}$ of the CH_3^+ fragments are given.

and one from HOMO–1 or HOMO–2 is a likely explanation. Since the asymmetry feature in CH_3^+ is only ~ 1 eV broad, the corresponding feature in the I^{q+} fragment spectrum would be 0.12 eV wide, well below the width of the Gaussian filter used to smooth the data. This fact, dissociation into more than two fragments or overlap with a dominant channel, may explain the absence of the expected asymmetry pattern in I^+ .

(3) KER 11 eV: the third horizontal line at 11 eV in panel (a) corresponds to CH_3^+ and I^{q+} with 9.8 and 1.2 eV kinetic energy, respectively. There is a clear peak at 11 eV in panel (c) showing a very weak asymmetry in I^{2+} , which has the same rather than the opposite sign as CH_3^+ . The asymmetry of I^{2+} [panel (c)] changes sign just above 11 eV. The observed asymmetry would also be compatible with that observed in I^+ [panel (b)], possibly as a consequence of sequential ionization. However, the high kinetic energy is incompatible with this assignment (see Table 3 for $q = 1$). Because of the close agreement with $E_{\text{CH}_3^+}^{\text{ai}} = 10.1$ eV, we assign this channel to $\text{CH}_3\text{I}^{3+} \rightarrow \text{CH}_3^+ + \text{I}^{2+}$, despite the lack of agreement in the emission asymmetry of the two cofragments.

(4) KER 14 eV: the horizontal green line at KER 14 eV corresponds to a kinetic energy of 12.6 eV for CH_3^+ and 1.4 eV for I^{q+} . The asymmetry in the ion yields is clearly opposite for I^{3+} in panel (d). The kinetic energy values match very well with ab initio calculations for $q = 3$ in Table 3. We unambiguously assign this channel to $\text{CH}_3\text{I}^{4+} \rightarrow \text{CH}_3^+ + \text{I}^{3+}$.

DISCUSSION

The asymmetries observed in the present experiments can originate from multiple sources. Since the electronic ground state of all CH_3X^+ molecules is strongly bound, single ionization from the HOMO will not lead to direct dissociation. However, CH_3X^+ in its electronic ground state may dissociate following absorption of photons from the two-color pulse, including bond softening, or excitation by the recolliding electron. These processes may result in asymmetric emission of low-energy charged fragments which is, however, not directly related to the asymmetry of the ionization process. Similarly, the electronic ground states of CH_3X^{2+} are bound and sufficiently long-lived to appear in the time-of-flight spectra (see also refs 44 and 46). However, in contrast to CH_3X^+ , the low barrier to Coulomb explosion (≈ 0.1 eV in $\text{CH}_3\text{I}^{2+42}$) and the pronounced shortening of the C–X bond upon double ionization from the antibonding HOMO will lead to direct dissociation of a large fraction of the molecules.

In addition to these mechanisms, ionization to electronically excited states of the cations can occur. Since the $\tilde{\text{A}}^+$ state of all CH_3X^+ molecules is weakly bound, a large fraction of molecules ionized out of HOMO–1 will dissociate directly. The same principle also applies to double ionization which can lead to electronically excited states of the dication (e.g., through ionization from HOMO and HOMO–1 or even more strongly bound orbitals).

The channel that offers the simplest interpretation with respect to the asymmetry of SFI is therefore the Coulomb

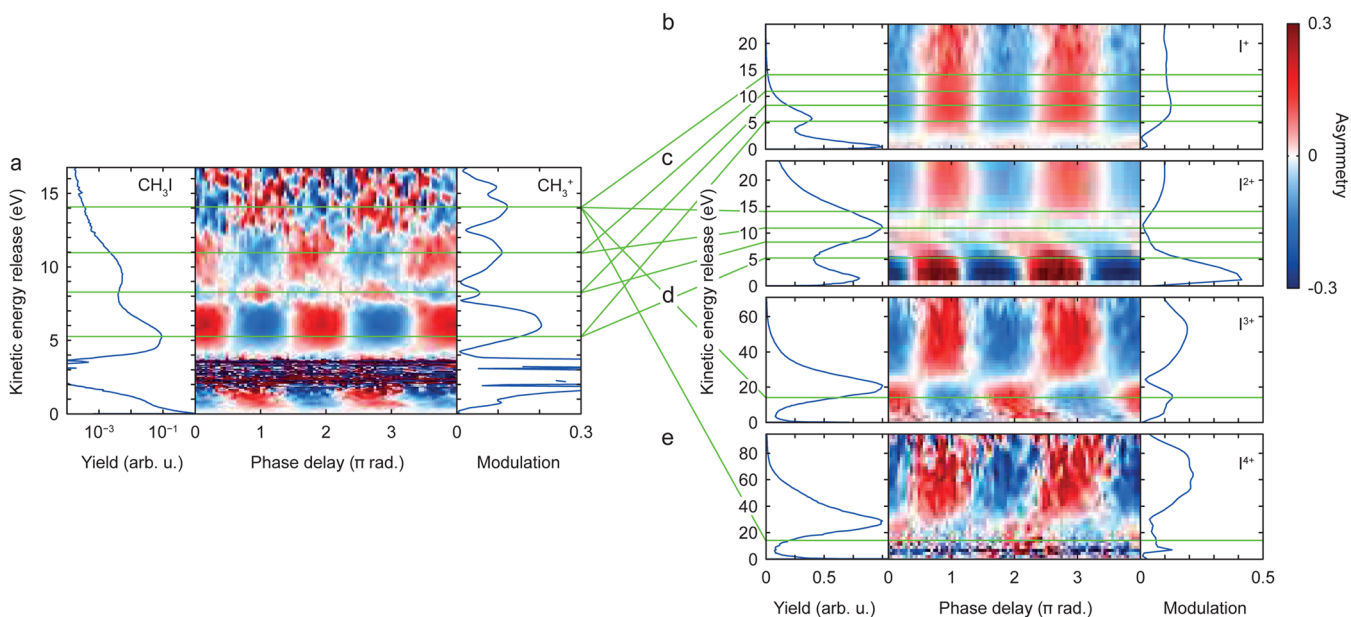


Figure 5. Correlation between different fragments from CH_3I : (a) CH_3^+ , (b) I^+ , (c) I^{2+} , (d) I^{3+} , and (e) I^{4+} . The laser peak intensity was $(1 \pm 0.4) \times 10^{14} \text{ W/cm}^2$. The measured abundance ratio of $\text{I}^+ : \text{I}^{2+} : \text{I}^{3+} : \text{I}^{4+}$ is 1:0.47:0.10:0.01. The green lines in (a–c) are plotted at KER values of 5.3, 8.3, 11, and 14 eV corresponding to local maxima in the modulation of the asymmetry yield, whereas the one in (d) and (e) corresponds to 14 eV.

explosion of the electronic ground state of $\text{CH}_3\text{X}^{2+} \rightarrow \text{CH}_3^+ + \text{X}^+$. This channel has been identified in all experiments in the range of CH_3^+ kinetic energies of 3.5–5 eV. Methyl fragments with less than 4.2 eV kinetic energy show opposite emission asymmetries in the case of CH_3F compared to CH_3Cl , CH_3Br , or CH_3I (Figure 2, panels a–d). As we show now, this observation can be explained by the asymmetry of SFI under the assumption that both electrons are removed from the HOMO. The orientation-dependent ionization rates were calculated based on the WFAT.^{17,18}

WFAT Ionization Rates. In the WFAT, the total ionization rate is given by

$$\Gamma = \sum_{n_\xi=0}^{\infty} \sum_{m=-\infty}^{\infty} \Gamma_{n_\xi m} + O(\Gamma^2) \quad (7)$$

where

$$\Gamma_{n_\xi m} = |G_{n_\xi m}(\beta, \gamma)|^2 W_{n_\xi m}(F) [1 + O(F)] \quad (8)$$

is the partial rate for ionization into a channel with parabolic quantum numbers n_ξ and m . In the leading-order approximation of the asymptotic expansion in field F , the ionization rate is determined by the dominant ionization channel with $n_\xi = m = 0$ and given by

$$\Gamma(\beta, \gamma) \approx \Gamma_{00} = |G_{00}(\beta, \gamma)|^2 W_{00}(F) [1 + O(F)] \quad (9)$$

where $W_{00}(F)$ is the field factor, an exponential function of the external field F .¹⁷ The orientation dependence of $\Gamma(\beta, \gamma)$ in this approximation thus does not depend on the field strength and is fully described by the structure factor $G_{00}(\beta, \gamma)$, where β and γ represent the Euler angles. The angle β is defined such that for $\beta = 0^\circ$, the electron is removed from the methyl side. The normalized values of the squared structure factor $|G_{00}(\beta, \gamma)|^2$ integrated over the angle γ are shown in Figure 6. The green and blue lines represent the results obtained for the two eigenfunctions Ψ_A and Ψ_B of the HOMO in a static electric field, respectively. These eigenfunctions are orientation-

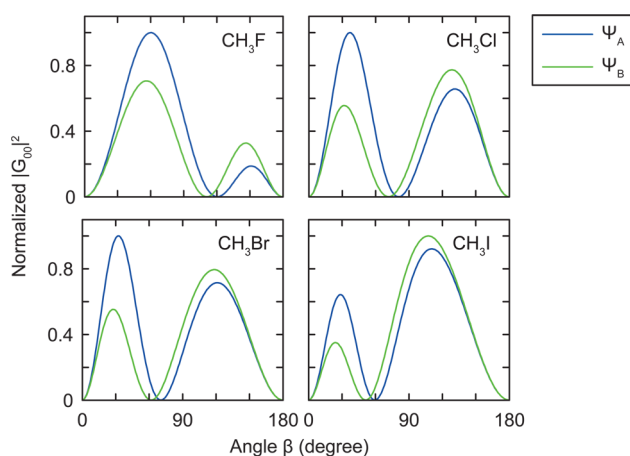


Figure 6. Calculated ionization rates of the HOMO. Angle-dependent structure factor $|G_{00}(\beta, \gamma)|^2$ integrated over γ for the methyl halides. $|G_{00}(\beta, \gamma)|^2$ was calculated for the two degenerate eigenfunctions Ψ_A and Ψ_B individually. For $\beta = 0^\circ$, the electron is removed from the methyl side.

dependent linear combinations of the field-free orbitals ϕ_a and ϕ_b ,⁴⁷ i.e.,

$$\begin{aligned} \Psi_A &= (-S\phi_a + C\phi_b) \text{sgn } C \\ \Psi_B &= (+C\phi_a + S\phi_b) \text{sgn } S \end{aligned} \quad (10)$$

where $S = \sin(\gamma/2 + \pi/4)$ and $C = \cos(\gamma/2 + \pi/4)$. The corresponding eigenvalues

$$z_{A/B} = (\mp U \sin \beta - V \cos \beta) \quad (11)$$

include the coupling constants U and V representing the linear Stark effect. The structure factor $G_{00}(\beta, \gamma)$ is obtained as explained in ref 47 using Hartree–Fock calculations. We use uncontracted polarization-consistent basis sets on the quintuple zeta (pc-4) quality level and variationally optimize all exponents of the basis set. Because the HOMO of the methyl halides has a

nodal plane containing the CX bond, $|G_{00}(\beta, \gamma)|^2$ is zero for $\beta = 0^\circ$ and $\beta = 180^\circ$. We now define the head-to-tail ionization asymmetry a as

$$a = \left(\int_0^{\pi/2} \int_0^{2\pi} \Gamma(\beta, \gamma) \sin(\beta) d\beta d\gamma - \int_{\pi/2}^{\pi} \int_0^{2\pi} \Gamma(\beta, \gamma) \sin(\beta) d\beta d\gamma \right) / \int_0^{\pi} \int_0^{2\pi} \Gamma(\beta, \gamma) \sin(\beta) d\beta d\gamma \quad (12)$$

The ionization asymmetries a for the two eigenfunctions of the HOMO are given in Table 4 and Figure 6. For $a > 0$, the

Table 4. Head-to-Tail Ionization Asymmetries a' Calculated within the Single-Active-Electron Version of the WFAT for the Eigenfunctions Ψ_A and Ψ_B of the HOMO in a Static Field^a

	a'_A	a'_B	a_A	a_B
CH ₃ F	0.75	0.68	0.69	0.58
CH ₃ Cl	0.66	0.28	-0.01	-0.45
CH ₃ Br	0.23	-0.12	-0.20	-0.46
CH ₃ I	-0.29	-0.35	-0.39	-0.43

^aThe asymmetries a were calculated within the WFAT, but the Stark shifts were calculated with non-perturbative multi-electron methods (see text for further details).

molecule preferentially ionizes from the methyl end. In this article, we compare results of the single-active-electron version of the WFAT^{17,18} where U and V are obtained as expectation values over ϕ_a and ϕ_b , which yields the asymmetries a' with results where U and V were obtained from multielectron quantum-chemical calculations with applied static electric fields yielding the asymmetries a . In more detail, the Stark shift of the binding energies was calculated on the HF/aug-cc-pVQZ level of theory for CH₃F, CH₃Cl, and CH₃Br and the HF/6-311G** level for CH₃I. The binding energies of the two components of the degenerate HOMO and HOMO-2 were calculated in the presence of static fields up to 0.05 au, and a second-order polynomial fit was used to extract the coefficients of the linear Stark effect. The U and V coefficients were determined by applying static fields along the three principal axes of the molecule and averaging over fields applied along $\pm z$ (V) or $\pm x$ and $\pm y$ (U). This approach includes multielectron effects into the treatment of the Stark effect and is consistent with the many-electron WFAT developed in ref 48. In the case of the nondegenerate HOMO-1 (a_1 symmetry), $U = 0$ by symmetry and V is obtained as described above. The HOMO-2 of all CH₃X molecules are degenerate (e symmetry) and were treated in the same way as the HOMO. All ionization-rate asymmetries provided by the WFAT are independent of the field strength within the leading-order approximation used in the present work.

The asymmetries a' , calculated within the single-active-electron approximation, predict that CH₃F, CH₃Cl, and CH₃Br all preferentially ionize via the methyl group, whereas this preference only reverts for CH₃I. This prediction contrasts with the measured asymmetry α (eq 4) of the CH₃X²⁺ \rightarrow CH₃⁺ + X⁺ channel generating low-energy CH₃⁺ fragments. Indeed, CH₃F is the only molecule of this series that preferentially ionizes via the methyl group. In contrast, the asymmetries a correctly

predict the experimentally observed asymmetries for all molecules (i.e., the fact that CH₃Cl, CH₃Br, and CH₃I all preferentially ionize via the halogen atom).

Comparing the modulation amplitude of the emission asymmetry shown in the right panels of Figure 2 (panels a–d) for fragment energies below 4 eV, one can conclude that the absolute value of the ionization asymmetry of CH₃F is higher compared to the other methyl halides. This aspect is also correctly predicted by the WFAT asymmetries a . The good agreement between the calculated ionization asymmetry and the emission asymmetry of the CH₃⁺ fragments confirms the assumption that the ionization asymmetry is the decisive quantity defining the asymmetry of the low-energy fragments from the CH₃X²⁺ \rightarrow CH₃⁺ + X⁺ Coulomb explosion of the molecules.

As mentioned above, the observed reversals of the emission asymmetry of high-energy CH₃⁺ ions from CH₃Cl, CH₃Br, and CH₃I might be attributed to ionization from lower-lying orbitals. We now investigate this hypothesis by studying the ionization asymmetries of HOMO-1 and HOMO-2. A selection of field-free valence orbitals of the four methyl halides are shown in Figure 7. The asymmetries a predict that CH₃F

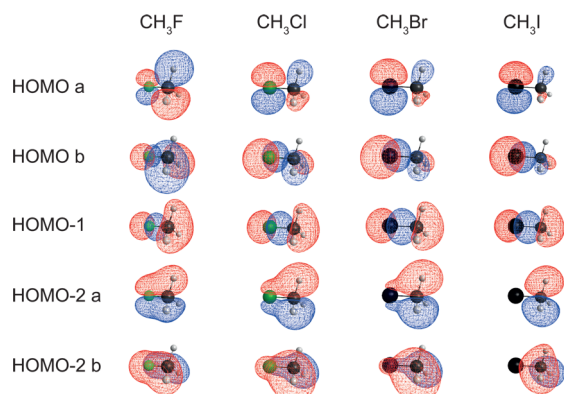


Figure 7. Isocountour representations of the field-free HOMO, HOMO-1, and HOMO-2 of the methyl halides. The color indicates the sign of the wave function. The HOMO and HOMO-2 are doubly degenerate and are labeled with a and b, according to eq 10. The orbital wave functions were calculated using the quantum chemistry program GAMESS (version 11; method: restricted HF; basis set: aug-ccpVQZ for CH₃F, CH₃Cl, and CH₃Br; Def2-QZVP for CH₃I).

ionizes faster if the electron is removed via the methyl side (i.e., the electric field pointing from CH₃ to F), where the electron density of the HOMO is maximal (see Figure 7). In contrast CH₃Cl, CH₃Br, and CH₃I preferentially ionize from the halogen atom, in agreement with the dominant amplitude of the HOMO. Hence, the orbital structure is found to dominate over the Stark effect in the case of HOMO. Considering tunneling ionization from HOMO-1 of the methyl halides (Figure 7), a smaller ionization asymmetry is expected to be observed due to the symmetric shape of the orbital. Indeed, the calculated ionization asymmetries a for HOMO-1 given in Table 5 are smaller compared to those of the HOMO. The calculations predict positive ionization asymmetries a for HOMO-1 of all 4 molecules. For HOMO-2, positive ionization asymmetries are obtained for both eigenfunctions (see Table 6) except for CH₃Cl for which a_B is negative but close to zero, whereas $a_A = 0.20$. Hence, one can conclude that for all methyl halides, ionization from HOMO-2 is enhanced if

Table 5. Similar to Table 4 but for HOMO–1

	a'	a
CH ₃ F	0.85	0.22
CH ₃ Cl	0.55	0.13
CH ₃ Br	0.28	0.50
CH ₃ I	–0.07	0.05

Table 6. Similar to Table 4 but for HOMO–2

	a'_A	a'_B	a_A	a_B
CH ₃ F	0.75	0.63	0.35	0.12
CH ₃ Cl	–0.64	–0.72	0.20	–0.04
CH ₃ Br	–0.81	–0.85	0.30	0.06
CH ₃ I	–0.91	–0.93	0.49	0.25

the electron is removed via the methyl end. In accordance with the shape of the field-free HOMO–2, this is again expected because the largest orbital amplitude is at the methyl side.

Our calculations thus predict that the sign of the ionization asymmetry of CH₃F is identical for ionization from the HOMO, HOMO–1, and HOMO–2. This is consistent with the absence of reversals in the emission asymmetry for CH₃⁺ ions from CH₃F. For CH₃Cl, CH₃Br, and CH₃I, where reversals of the emission asymmetry were observed, the sign of the calculated ionization asymmetries obtained for the HOMO are different from the asymmetries obtained for lower-lying orbitals. Therefore, it is likely that the changes of the emission asymmetry of the high-energy CH₃⁺ fragments are due to ionization from low-lying orbitals,^{49–51} leading to electronically excited states of CH₃X²⁺ and CH₃X³⁺. In this context, the asymmetry in CH₃⁺ from CH₃I at 8.3 eV (see Figure 5) may be assigned to ionization from HOMO and HOMO–2, which has a similar but opposite ionization asymmetry as compared to HOMO. HOMO–1, in contrast, has too small an ionization asymmetry to overturn the asymmetry imposed by HOMO. Ionization from HOMO–2 could occur sequentially to ionization from HOMO or through recollision. An improved theoretical analysis of these results will require the treatment of sequential ionization, electron recollision, and nuclear dynamics.

CONCLUSION

We have measured the asymmetric emission of charged fragments from methyl halides in intense phase-controlled two-color fields for a range of laser intensities. From the emission asymmetry of the low-energy CH₃⁺ fragments measured at moderate laser intensities, we conclude that SFI of CH₃F maximizes when the electric field points from the methyl group to the halogen atom, whereas for CH₃Cl, CH₃Br, and CH₃I SFI preferentially proceeds in the opposite direction. We attribute this result to the asymmetric ionization rate of the HOMO of the neutral methyl halide molecules. This conclusion also agrees with the ionization rate predicted by the WFAT, provided that the Stark effect is obtained from nonperturbative multielectron calculations with applied electric fields. Further, we have observed changes of the emission asymmetries of CH₃⁺ fragments from CH₃Cl, CH₃Br, and CH₃I as a function of the kinetic-energy release. We propose that these changes of the asymmetry result from ionization to electronically excited states of the doubly and triply charged parent molecules (e.g., by ionization from HOMO and HOMO–2). This assignment is supported by the calculated

ionization asymmetries for these orbitals. Further theoretical developments and modeling are required to fully establish this aspect.

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Notes

The authors declare no competing financial interest.

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REFERENCES

- Itatani, J.; Levesque, J.; Zeidler, D.; Niikura, H.; Pépin, H.; Kieffer, J. C.; Corkum, P. B.; Villeneuve, D. M. Tomographic Imaging of Molecular Orbitals. *Nature* **2004**, *432*, 867–871.
- Baker, S.; Robinson, J. S.; Haworth, C. A.; Teng, H.; Smith, R. A.; Chiril, C. C.; Lein, M.; Tisch, J. W. G.; Marangos, J. P. Probing Proton Dynamics in Molecules on an Attosecond Time Scale. *Science* **2006**, *312*, 424–427.
- Haessler, S.; Caillat, J.; Boutu, W.; Giovanetti-Teixeira, C.; Ruchon, T.; Auguste, T.; Diveki, Z.; Breger, P.; Maquet, A.; Carré, B.; et al. Attosecond Imaging of Molecular Electronic Wavepackets. *Nat. Phys.* **2010**, *6*, 200–206.
- Wörner, H. J.; Bertrand, J. B.; Kartashov, D. V.; Corkum, P. B.; Villeneuve, D. M. Following a Chemical Reaction using High-Harmonic Interferometry. *Nature* **2010**, *466*, 604–607.
- Wörner, H. J.; Bertrand, J. B.; Fabre, B.; Higué, J.; Ruf, H.; Dubrouil, A.; Patchkovskii, S.; Spanner, M.; Mairesse, Y.; Blanchet, V.; et al. Conical Intersection Dynamics in NO₂ Probed by Homodyne High-Harmonic Spectroscopy. *Science* **2011**, *334*, 208–212.
- Kraus, P. M.; Baykusheva, D.; Wörner, H. J. Two-Pulse Field-Free Orientation Reveals Anisotropy of Molecular Shape Resonance. *Phys. Rev. Lett.* **2014**, *113*, 023001.
- Okunishi, M.; Morishita, T.; Prümper, G.; Shimada, K.; Lin, C. D.; Watanabe, S.; Ueda, K. Experimental Retrieval of Target Structure Information from Laser-Induced Rescattered Photoelectron Momentum Distributions. *Phys. Rev. Lett.* **2008**, *100*, 143001.
- Ray, D.; Ulrich, B.; Bocharova, I.; Maharjan, C.; Ranitovic, P.; Gramkow, B.; Magrakvelidze, M.; De, S.; Litvinyuk, I. V.; Le, A. T.; et al. Large-Angle Electron Diffraction Structure in Laser-Induced Rescattering from Rare Gases. *Phys. Rev. Lett.* **2008**, *100*, 143002.
- Meckel, M.; Comtois, D.; Zeidler, D.; Pavicic, D.; Bandulet, H. C.; Pépin, H.; Kieffer, J. C.; Dorner, R.; Villeneuve, D. M.; Corkum, P. B.; et al. Laser-Induced Electron Tunneling and Diffraction. *Science* **2008**, *320*, 1478–1482.
- Blaga, C. I.; Xu, J.; DiChiara, A. D.; Sistrunk, E.; Zhang, K.; Agostini, P.; Miller, T. A.; DiMauro, L. F.; Lin, C. D. Imaging Ultrafast Molecular Dynamics with Laser-Induced Electron Diffraction. *Nature* **2012**, *483*, 194–197.
- Bian, X.-B.; Huismans, Y.; Smirnova, O.; Yuan, K.-J.; Vrakking, M. J. J.; Bandrauk, A. D. Subcycle Interference Dynamics of Time-Resolved Photoelectron Holography with Mid-Infrared Laser Pulses. *Phys. Rev. A: At, Mol., Opt. Phys.* **2011**, *84*, 043420.
- Hickstein, D. D.; Ranitovic, P.; Witte, S.; Tong, X.-M.; Huismans, Y.; Arpin, P.; Zhou, X.; Keister, K. E.; Hogle, C. W.; Zhang, B.; et al. Direct Visualization of Laser-Driven Electron Multiple

Scattering and Tunneling Distance in Strong-Field Ionization. *Phys. Rev. Lett.* **2012**, *109*, 073004.

(13) Meckel, M.; Staudte, A.; Patchkovskii, S.; Villeneuve, D. M.; Corkum, P. B.; Dörner, R.; Spanner, M. Signatures of the Continuum Electron Phase in Molecular Strong-Field Photoelectron Holography. *Nat. Phys.* **2014**, *10*, 594–600.

(14) Tong, X. M.; Zhao, Z. X.; Lin, C. D. Theory of Molecular Tunneling Ionization. *Phys. Rev. A: At., Mol., Opt. Phys.* **2002**, *66*, 033402.

(15) Muth-Böhm, J.; Becker, A.; Faisal, F. H. M. Suppressed Molecular Ionization for a Class of Diatomics in Intense Femtosecond Laser Fields. *Phys. Rev. Lett.* **2000**, *85*, 2280.

(16) Computational Chemistry Comparison and Benchmark Database -Release 17b, NIST. <http://cccbdb.nist.gov> (accessed November 2015).

(17) Tolstikhin, O. I.; Morishita, T.; Madsen, L. B. Theory of Tunneling Ionization of Molecules: Weak-Field Asymptotics Including Dipole Effects. *Phys. Rev. A: At., Mol., Opt. Phys.* **2011**, *84*, 053423.

(18) Madsen, L. B.; Tolstikhin, O. I.; Morishita, T. Application of the Weak-Field Asymptotic Theory to the Analysis of Tunneling Ionization of Linear Molecules. *Phys. Rev. A: At., Mol., Opt. Phys.* **2012**, *85*, 053404.

(19) Trinh, V. H.; Tolstikhin, O. I.; Madsen, L. B.; Morishita, T. First-Order Correction Terms in the Weak-Field Asymptotic Theory of Tunneling Ionization. *Phys. Rev. A: At., Mol., Opt. Phys.* **2013**, *87*, 043426.

(20) Trinh, V. H.; Pham, V. N. T.; Tolstikhin, O. I.; Morishita, T. Weak-Field Asymptotic Theory of Tunneling Ionization Including the First-Order Correction Terms: Application to Molecules. *Phys. Rev. A: At., Mol., Opt. Phys.* **2015**, *91*, 063410.

(21) Otobe, T.; Yabana, K.; Iwata, J.-I. First-Principles Calculations for the Tunnel Ionization Rate of Atoms and Molecules. *Phys. Rev. A: At., Mol., Opt. Phys.* **2004**, *69*, 053404.

(22) Fowe, E. P.; Bandrauk, A. D. Nonperturbative Time-Dependent Density-Functional Theory of Ionization and Harmonic Generation in OCS and CS₂ Molecules with Ultrashort Intense Laser Pulses: Intensity and Orientational Effects. *Phys. Rev. A: At., Mol., Opt. Phys.* **2011**, *84*, 035402.

(23) Zhang, B.; Yuan, J.; Zhao, Z. Dynamic Core Polarization in Strong-Field Ionization of CO Molecules. *Phys. Rev. Lett.* **2013**, *111*, 163001.

(24) Petretti, S.; Vanne, Y. V.; Saenz, A.; Castro, A.; Decleva, P. Alignment-Dependent Ionization of N₂, O₂, and CO₂ in Intense Laser Fields. *Phys. Rev. Lett.* **2010**, *104*, 223001.

(25) Majety, V. P.; Zielinski, A.; Scrinzi, A. Photoionization of Few Electron Systems: A Hybrid Coupled Channels Approach. *New J. Phys.* **2015**, *17*, 063002.

(26) Holmegaard, L.; Hansen, J. L.; Kalhj, L.; Kragh, S. L.; Stapelfeldt, H.; Filsinger, F.; Küpper, J.; Meijer, G.; Dimitrovski, D.; Abu-samha, M.; et al. Photoelectron Angular Distributions from Strong-Field Ionization of Oriented Molecules. *Nat. Phys.* **2010**, *6*, 428–432.

(27) Dimitrovski, D.; Martiny, C. P. J.; Madsen, L. B. Strong-Field Ionization of Polar Molecules: Stark-Shift-Corrected Strong-Field Approximation. *Phys. Rev. A: At., Mol., Opt. Phys.* **2010**, *82*, 053404.

(28) Dimitrovski, D.; Abu-samha, M.; Madsen, L. B.; Filsinger, F.; Meijer, G.; Küpper, J.; Holmegaard, L.; Kalhj, L.; Nielsen, J. H.; Stapelfeldt, H. Ionization of Oriented Carbonyl Sulfide Molecules by Intense Circularly Polarized Laser Pulses. *Phys. Rev. A: At., Mol., Opt. Phys.* **2011**, *83*, 023405.

(29) Hansen, J. L.; Holmegaard, L.; Nielsen, J. H.; Stapelfeldt, H.; Dimitrovski, D.; Madsen, L. B. Orientation-Dependent Ionization Yields from Strong-Field Ionization of Fixed-in-Space Linear and Asymmetric Top Molecules. *J. Phys. B: At., Mol. Opt. Phys.* **2012**, *45*, 015101.

(30) Kraus, P. M.; Rupenyan, A.; Wörner, H. J. High-Harmonic Spectroscopy of Oriented OCS Molecules: Emission of Even and Odd Harmonics. *Phys. Rev. Lett.* **2012**, *109*, 233903.

(31) Ohmura, H.; Saito, N.; Morishita, T. Molecular Tunneling Ionization of the Carbonyl Sulfide Molecule by Double-Frequency Phase-Controlled Laser Fields. *Phys. Rev. A: At., Mol., Opt. Phys.* **2014**, *89*, 013405.

(32) Madsen, L. B.; Jensen, F.; Tolstikhin, O. I.; Morishita, T. Structure Factors for Tunneling Ionization Rates of Molecules. *Phys. Rev. A: At., Mol., Opt. Phys.* **2013**, *87*, 013406.

(33) Ohmura, H.; Ito, F.; Tachiya, M. Phase-Sensitive Molecular Ionization Induced by a Phase-Controlled Two-Color Laser Field in Methyl Halides. *Phys. Rev. A: At., Mol., Opt. Phys.* **2006**, *74*, 043410.

(34) Thompson, M. R.; Thomas, M. K.; Taday, P. F.; Posthumus, J. H.; Langley, A. J.; Frasiniski, L. J.; Codling, K. One and Two-Color studies of the Dissociative Ionization and Coulomb Explosion of H₂ with Intense Ti:Sapphire Laser Pulses. *J. Phys. B: At., Mol. Opt. Phys.* **1997**, *30*, S755.

(35) Ray, D.; He, F.; De, S.; Cao, W.; Mashiko, H.; Ranitovic, P.; Singh, K. P.; Znakovskaya, I.; Thumm, U.; Paulus, G. G.; et al. Ion-Energy Dependence of Asymmetric Dissociation of D₂ by a Two-Color Laser Field. *Phys. Rev. Lett.* **2009**, *103*, 223201.

(36) Sheehy, B.; Walker, B.; DiMauro, L. F. Phase Control in the Two-Color Photodissociation of HD⁺. *Phys. Rev. Lett.* **1995**, *74*, 4799.

(37) Li, H.; Rod, D.; De, S.; Znakovskaya, I.; Cao, W.; Laurent, G.; Wang, Z.; Kling, M. F.; Le, A. T.; Cocke, C. L. Orientation Dependence of the Ionization of CO and NO in an Intense Femtosecond Two-Color Laser Field. *Phys. Rev. A: At., Mol., Opt. Phys.* **2011**, *84*, 043429.

(38) Ohmura, H.; Saito, N.; Tachiya, M. Selective Ionization of Oriented Nonpolar Molecules with Asymmetric Structure by Phase-Controlled Two-Color Laser Fields. *Phys. Rev. Lett.* **2006**, *96*, 173001.

(39) Ohmura, H.; Saito, N.; Nonaka, H.; Ichimura, S. Dissociative Ionization of a Large Molecule Studied by Intense Phase-Controlled Laser Fields. *Phys. Rev. A: At., Mol., Opt. Phys.* **2008**, *77*, 053405.

(40) Walt, S. G. Imaging Electronic Structure and Dynamics of Molecules through Strong-Field Ionization, Rescattering and Holography. Ph.D. Thesis, ETH Zurich, 2015.

(41) Vrakking, M. J. J. An Iterative Procedure for the Inversion of Two-Dimensional Ion/Photoelectron Imaging Experiments. *Rev. Sci. Instrum.* **2001**, *72*, 4084–4089.

(42) Corrales, M. E.; Gitzinger, G.; González-Vázquez, J.; Lorient, V.; de Nalda, R.; Banares, L. Velocity Map Imaging and Theoretical Study of the Coulomb Explosion of CH₃I under Intense Femtosecond IR Pulses. *J. Phys. Chem. A* **2012**, *116*, 2669–2677.

(43) Liu, H.; Yang, Z.; Gao, Z.; Tang, Z. Ionization and Dissociation of CH₃I in Intense Laser Field. *J. Chem. Phys.* **2007**, *126*, 044316.

(44) Sun, S.; Yang, Y.; Zhang, J.; Wu, H.; Chen, Y.; Zhang, S.; Jia, T.; Wang, Z.; Sun, Z. Ejection of Triatomic Molecular Ion H₃⁺ Image from Methyl Chloride in an Intense Femtosecond Laser Field. *Chem. Phys. Lett.* **2013**, *581*, 16–20.

(45) Kotsina, N.; Kaziannis, S.; Kosmidis, C. Hydrogen Migration in Methanol Studied under Asymmetric fs Laser Irradiation. *Chem. Phys. Lett.* **2014**, *604*, 27–32.

(46) Tanaka, M.; Murakami, M.; Yatsushashi, T.; Nakashima, N. Atomic-like Ionization and Fragmentation of a Series of CH₃X (X: H, F, Cl, Br, I, and CN) by an Intense Femtosecond Laser. *J. Chem. Phys.* **2007**, *127*, 104314.

(47) Kraus, P. M.; Tolstikhin, O. I.; Baykusheva, D.; Rupenyan, A.; Schneider, J.; Bisgaard, C. Z.; Morishita, T.; Jensen, F.; Madsen, L. B.; Wörner, H. J. Observation of Laser-Induced Electronic Structure in Oriented Polyatomic Molecules. *Nat. Commun.* **2015**, *6*, 7039.

(48) Tolstikhin, O. I.; Madsen, L. B.; Morishita, T. Weak-Field Asymptotic Theory of Tunneling Ionization in Many-Electron Atomic and Molecular Systems. *Phys. Rev. A: At., Mol., Opt. Phys.* **2014**, *89*, 013421.

(49) Akagi, H.; Otobe, T.; Staudte, A.; Shiner, A.; Turner, F.; Dörner, R.; Villeneuve, D. M.; Corkum, P. B. Laser Tunnel Ionization from Multiple Orbitals in HCl. *Science* **2009**, *325*, 1364–1367.

(50) Wu, J.; Schmidt, L. P. H.; Kunitzki, M.; Meckel, M.; Voss, S.; Sann, H.; Kim, H.; Jahnke, T.; Czasch, A.; Dörner, R. Multiorbital

Tunneling Ionization of the CO Molecule. *Phys. Rev. Lett.* **2012**, *108*, 183001.

(51) Liu, H.; Zhao, S.-F.; Li, M.; Deng, Y.; Wu, C.; Zhou, X.-X.; Gong, Q.; Liu, Y. Molecular-Frame Photoelectron Angular Distributions of Strong-Field Tunneling from Inner Orbitals. *Phys. Rev. A: At, Mol, Opt. Phys.* **2013**, *88*, 061401.